Floating between two liquids

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Abstract

The hydrostatic equilibrium of a solid floating “between two liquids” is analysed, first as a classical exercise, then as a simple experiment with a particular staging. It is suggested to use this situation to address some students’ difficulties about: pressure in statics of fluids, role of atmospheric pressure, Archimedes’ up-thrust, and thinking this setting as a system. It is shown that a possible way to help students overcome some of these difficulties, especially the last one, is to use graphical representations.

A very classical exercise is the analysis of the hydrostatic equilibrium of a cylindrical body – height $H$, area of the base $S$, mean density $\rho_s$ - situated in a recipient with two immiscible liquids of density, respectively $\rho_1$ and $\rho_2$, such that $\rho_1 > \rho_s > \rho_2$. The body floats in the first one alone ($\rho_1 > \rho_s$) and sinks in the other one alone ($\rho_s > \rho_2$). The hydrostatic equilibrium of the solid “between two liquids” may be like one of the two cases ($a$ or $b$) represented in Figure 1.

Figure 1 Two cases for a cylindrical solid at hydrostatic equilibrium “between” two liquids.
Materials

The materials needed to implement this experiment are very simple, for example.
- A transparent cylindrical glass filled with water (possibly with a colorant).
- Groundnut oil, usually about 50 cm³ are sufficient.
- A plastic egg (toys, e.g. Kinder) or any empty tube of pills, to be partly filled with dense materials (e.g. coins, shot) until the floating in water (alone) is ensured. It is in a stable vertical position, the top of the solid emerging by about the third or the quarter of its height $H$ out the free surface of water.

Classical solution

For any position of the solid, the relationship of fluid statics $\Delta p = -\rho g \Delta z$ can be used for each part of the cylinder immersed in each liquid, of respective heights $h_1$ and $h_2$. Therefore, with an upward axis, the differences in pressure between lower and upper horizontal sections of the cylinder immersed in, respectively, liquid 1 and 2 are given by:

$\Delta p_1 = \rho_1 g h_1$

$\Delta p_2 = \rho_2 g h_2$

The possible contribution of the air (case b in Fig. 1) can be neglected with respect to the two others, given that the density of the air is typically a thousand times smaller than those of the liquids.

A state of equilibrium occurs when Archimedes’ up-thrusts due to each liquid and the weight of the body balance out:

$(\rho_1 h_1 + \rho_2 h_2) S g - \rho_s H S g = 0$  or else:

$\rho_1 h_1 + \rho_2 h_2 = \rho_s H$

Solving for, say, $h_1$ gives

$h_1 = (\rho_s H - \rho_2 h_2)/\rho_1$  (1)

$h_1$ and $h_2$ being the heights of each part of the cylinder immersed in each liquid in case the cylinder actually floats between the liquids.

In particular, relationship (1) gives the value of $h_1$ that is necessary for the body to float in liquid 1 alone (then $h_2 = 0$), or to be in a state of equilibrium of the type “floating between two liquids” with a given value of $h_2$. Would the actual volume of liquid 1 be too small to ensure this condition on $h_1$, then the body would rest on the bottom of the recipient.

The statements of this section, and in particular relationship (1), hold for case a and case b in Figure 1; with, in case b, the approximation that Archimedes’ up-thrust due to air is neglected with respects to the two other contributions.

In case a, when the body is covered by fluid 2, $h_1 + h_2 = H$ and relationship (1) leads to

$h_1 = h_2 (\rho_1 - \rho_3) / (\rho_1 - \rho_s)$  (2)

or else

$h_1 = H (\rho_1 - \rho_3) / (\rho_1 - \rho_2)$  (3)

Stability of the equilibrium “between two fluids”: For the cylindrical solid to stay in a stable vertical position, it is necessary that the center of mass of the solid be lower than the center of
mass of a fluid cylinder of same section filled up with the two liquids respectively by a height $h_1$ and $h_2$. The insertion in the solid cylinder of a dense material, e.g. coins or shot, can solve this problem.

Practical detail: when the immersed cylinder is just resting (with nearly zero interaction) on the bottom of the recipient, it is particularly striking to see it “taking off” when the oil is added.

This situation can be used to
- address some students’ difficulties about: pressure in statics of fluids, role of atmospheric pressure, Archimedes’ up-thrust, thinking this setting as a system.
- overcome some of these difficulties, especially the last one, by using graphical representations, thereby addressing the interpretation of abstract representations as Cartesian graphs.

Evidencing possible difficulties

A questionnaire by Bennhold & Feldmann (2005) shows that the following question raises difficulties in students:

**Question:** A cylindrical body (mean density $\rho_s$) floats on a liquid 1 (density $\rho_1$). Another liquid (2: $\rho_2 < \rho_s$) is poured on top of liquid 1. The two liquids are not miscible. What will happen with the cylinder? Choose the right answer and explain: The new equilibrium position of the cylinder will be

A. same as before  
B. higher than before  
C. lower than before  
D. at the bottom  
E. more information is needed

(In Bennhold & Feldmann ‘s wording, it is said that “an object floats in water with ¾ of it’s volume submerged, and “oil is poured on top of the water”; and the questions are slightly different).

**A correct answer**

Adding liquid 2 results in a part of the solid being then immersed in this liquid, with a corresponding contribution to Archimedes’ up-thrust. Relationship (1) shows that a non-zero value of $h_2$ always contributes to a smaller value of $h_1$, comparing the equilibrium “between two fluids” to the situation when the solid floats in liquid 1 only. The solid will then move up.

However, $h_2$ being always smaller than $H$ (or equal to $H$), and $\rho_2$ being smaller than $\rho_s$ (the body is said not to float in liquid 2), relationship (1) also entails that $h_1$ cannot be zero. A part of the cylinder will stay in liquid 1.

**The wrong answers observed by Bennhold & Feldmann** are nearly exclusively A) and C), and, in all, concern about 75% of their sample of students.
**Possible origins of common difficulties**

Common answers may be due to

- the idea that a liquid in which the solid cannot float \( \rho_2 < \rho_s \) cannot facilitate the floating of this solid - floating which is ensured only by the first liquid (item A).

- the idea of a load added on top of the cylinder, pushing downwards (items C and D). It is likely that this idea will be particularly strong when it is stated that the solid, before being released from its initial position, is covered by liquid 2. This common idea contradicts a well-known fact: If you add water in a recipient with an object floating in water, this object will float to the top. But seeing a contradiction is not enough to reach a satisfying comprehension.

In both cases, the change in the setting due to the addition of liquid 2 is not envisaged in a systemic way, but **very locally**. Instead of taking into account that the whole system experiences changes (here in pressure), many students seem to consider that only a local change occurs.

This tendency has been observed since long, in particular in mechanics and fluid statics (Fauconnet 1981), electric circuits (Shipstone 1985, Closset 1985) and thermodynamics (Rozier & Viennot 1991). For a synthesis on this point, see Viennot (2001, chap. 5)

This suggests the following staging of the situation.

**Staging the situation in order to stress the systemic aspect, via graphs**

**Asking for a prediction with justification**

The question formulated in detail above can be asked (in brief: what will happen if some oil is poured into the glass?). In order to correctly answer this question, it is necessary that the students know the data provided above (relation between densities, the fact that fluids are immiscible) but it can be chosen to let students ask their own questions in this respect. Note that is it possible to specify – or not - that the cylinder will be covered by liquid 2 *before* being released from its initial position, or else to leave the students ask themselves this question.

The next steps consist of:

- helping students discuss about the reasons for their answer;
- letting them perform the experiment, discuss specific outcomes and possible discrepancies and similarities between the choices in the proposed answers;
- facilitating a further discussion in order to reach, if possible, a consensus.

*Practical detail:* when the immersed cylinder is just resting (with nearly zero interaction) on the bottom of the recipient, it is particularly striking to see it “taking off” when the oil is added.
**Staging the situation in order to stress the systemic aspect**

Among the facilitating tools that can be used with students, beyond the classical solution, the graphical approach is of interest.

A graph can be constructed, showing pressure against altitude without (black line in fig. 2) and with (coloured line in fig. 2) the second liquid.

The approximation done previously about the role of the air - i.e. a negligible contribution to Archimedes’ up-thrust - has a graphical equivalent: a constant value of atmospheric pressure near the recipient.

What counts to evaluate the up-thrust on the cylinder is the difference between the pressure forces exerted on the lower and upper horizontal surfaces of this body, situated at a given altitude interval $H$. For **any hydrostatic equilibrium**, this difference in pressure, $\Delta p_{eq}$, is such that

$$S \Delta p_{eq} = g (\rho_s S H)$$

or else

$$\Delta p_{eq} = g \rho_s H$$

![Graph showing pressure versus altitude in two situations: only liquid 1 (black), liquid 2 added on top of liquid 1 (coloured). When the second liquid is added the solid will move upwards (due to increased $\Delta p$), reaching a new equilibrium: see Fig. 3.](image)

It can be seen that, at the initial equilibrium position, the difference in pressure $\Delta p$ between the lower and the upper horizontal surface limiting the cylinder is larger when the second liquid is added. Hence the resulting increased up-thrust. The body then moves upwards and reaches a new equilibrium position, where $\Delta p$ retrieves its initial value (fig. 3).
An important point to be stressed is that the effect of adding the second fluid on top of the first one is a change of the whole field of pressure: a systemic view.

As regards Figure 2, some possibly useful questions to ask the students may be:
- explain, in your words, why, concerning a part of the liquid, the coloured and black lines are parallel;
- explain the physical meaning of their distance when these lines are parallel;
- explain what happens when the atmospheric pressure increases;
- explain what happens if the two liquids have the same density (the case when liquid 1 = liquid 2.

All the various cases (not enough water to ensure floating only in liquid 1, case a and b in Figure 1, density $\rho_2$ lower than $\rho_1$ and larger than $\rho_s$) can be discussed in terms of the difference in pressure $\Delta p$ between lower an upper sections of the solid.

Performing all the corresponding experiments after justified predictions will feed all the discussions about these various cases.

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**Fig. 3 Looking for the new equilibrium.** For given values of $H$ and $\Delta p_{eq}$, the “set square” in the top right angle should be moved and “stuck” against the coloured line representing the field of pressure. This gives at the same time the new position of the body and the new values of the heights of the immersed parts.

**Practical suggestion: the set square**

Finding any equilibrium position sums up in fitting:
- the field of pressure, graphically represented, and characteristic of the fluid.
- two characteristics of the body: height $H$ and $\Delta p_{eq}$ \hspace{1cm} ($\Delta p_{eq} = g \rho_s H$)
This can be highlighted by using a cardboard set-square and fitting it on a graph drawn on the blackboard, like in fig. 3. Many different situations can be solved with this technique.

Using the set square also makes it possible to predict qualitatively the force exerted by the fluids on the cylinder when this body is pushed downward by a distance \( d \) from its equilibrium position.

Note that the order of magnitude of atmospheric pressure, compared with that of pressure differences in this situation (x100), does not allow to set the origin of pressure axis at the intersection of the axes.

**Links with other situations**

▲ Stressing that adding the second fluid on top of the first one changes the *whole* field of pressure comes down to emphasize the need for a systemic analysis. This essential idea is also very useful to overcome common ideas such that: *pressure is directly linked to the weight of the column of fluid that is “above your head”*. For instance, it is easy to evidence the corresponding risks of misunderstanding with the two-fish situation (fig. 3, see Pugliese-Jona 1984), previously discussed in the paper “Various experiments involving fluid statics” (Viennot *et al.*, MUSE). As documented by Besson (Besson 2004, Besson & Viennot 2004, Viennot 2003), a large proportion of students tend to think that pressure is lower in an underwater cave (fish 1) than in the open sea at same depth (fish 2), as if the height of water above the cave’s ceiling affected pressure only in the open sea.

![Figure 3 Two-fish situation](image)

▲ The two-fish problem is also an opportunity to stress the fact that Archimedes’ up-thrust should be seen, first of all, as the outcome of differences in values of pressure in the fluid. All too often, the mere use of the formula screens this central aspect. Still more generally, the importance of differences in physics is so crucial that many other situations deserve the same spotlighting (see for instance “Staging the siphon” (Viennot & Planinsic 2009) on this MUSE site.
An activity with a slightly more complex situation – a Cartesian diver between two fluids – has been proposed and analysed by G. Planinsic, M. Kos and R. Jerman (2004). Then, the transformation envisaged is to squeeze the container and discuss how the state of equilibrium is changed. The main focus is then on the change of volume of the diver, but an analysis of the changes in pressure is also suggested, as well as the students’ difficulties in this respect.

References


